Synthesis of Chiral 5-Substituted 2-Pyrrolidinones: An Unusual One-Step Transformation

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Abstract. An efficient methodology for the enantioselective synthesis of γ -lactams, using a one-pot reaction of chiral N-alkoxycarbonyl γ -amino α , β -unsaturated carboxylates with magnesium in methanol, afforded the target chiral compounds in high chemical yield (87-95%) and optical purity (96-99% ee).

The 2-pyrrolidinone (γ -lactam) ring system is present in many biological active compounds and it is also a versatile intermediate particularly for the syntheses of novel amino acids and pyrrolidine derivatives.¹⁻⁴ Thus, it is not surprising that numerous racemic syntheses⁵⁻⁷ of these heterocycles have been described. Meyers *et al.*¹ recently described an efficient asymmetric synthesis of 5-substituted 2-pyrrolidinones that should find general applicability to a variety of synthetic challenges.





Magnesium-methanol is a simple, yet selective reagent, for the reduction of α,β -unsaturated esters.^{8,9} We have now found that the reaction of N-protected γ -amino- α,β -unsaturated carboxylates (1) with magnesium in methanol afforded 5-substituted 2-pyrrolidinones (2) in excellent yield (Scheme 1). Some examples that illustrate this reaction are summarized in Table 1. Transesterification occurred in all cases. In a typical procedure, a mixture of the N-protected γ -amino- α,β -unsaturated carboxylate (1, 1.0 mmol) and magnesium turnings (0.24 g, 10 mmol) in 10 mL methanol was stirred for 4 hr at 0°C and then for 8 hr at 25°C. After neutralization with hydrochloric acid (2N), the mixture was extracted with chloroform (2 x 40 mL). The combined chloroform extracts were washed with brine (10 mL), dried (MgSO₄), and the solvent was removed *in vacuo* to give a residue which was purified by silica gel column chromatography to yield the 5-substituted 2-pyrrolidinone (2).



Table 1. Synthesis of Chiral 5-Substituted 2-Pyrrolidinones¹⁰





The mechanism of this one-pot reaction may involve the formation of an intermediate N-protected γ amino carboxylate (3), since treatment of the (S)-N-protected γ -amino- α , β -unsaturated carboxylate (1g) with magnesium in CD₃OD afforded an approximately 1:1 mixture of the 5-substituted 2-pyrrolidinones 2d and 2e (Entry 8). It is plausible that cleavage of the carbamate and ester moieties and cyclization, take place during the formation of the 5-substituted γ -lactams (2) from the intermediate 3, via the transition states A, B and C (see Scheme 2). This explanation is based on the following observations. The carbamates 4a and 4b were not cleaved using the same reaction conditions. Secondly, reaction of the N-protected (S)-glutamic acid diethyl ester (5) with magnesium in methanol gave the pyroglutamic acid methyl ester (6, Equation 1). Finally, using the same reaction conditions, treatment of the N-diprotected γ -amino- α , β -unsaturated carboxylates (S)-1h and (S)-1i afforded the N-diprotected- γ -amino carboxylates 7a and 7b resulting from normal reduction of the α , β unsaturated olefinic bond.⁸ In these latter reactions, neither carbamate-cleavage, or γ -lactam, products were detected (Equations 2 and 3).



The optical purity of the γ -lactams (R)-2a, (R)-2b, (R)-2c and (S)-2c was determined by ¹H nmr analysis of the respective diastereomeric ureide prepared by reaction with (R)-1-phenylethyl isocyanate.¹³ The ¹H nmr spectrum of the (R)-2c and (S)-2c ureide derivatives showed ¹H resonances for MeS- as a singlet at δ 2.130 and 2.073, respectively. The ureide derivative of (R)-2a and (R)-2b existed as a single diastereomer since the ¹H nmr spectrum, upon CH-Me irradiation, showed that the benzylic methine proton appeared as a doublet (J_{win}=8 Hz) at δ 5.014 and 5.012, respectively.

The conversion of N-protected γ -amino- α , β -unsaturated carboxylates (1) to their corresponding 5substituted 2-pyrrolidinones (2) normally requires four reactions which include reduction of the C=C double bond, cleavage of the N-protected and ester moieties, and cyclization of the γ -amino acid. The one-pot transformation described provides an efficient procedure for the synthesis of chiral 5-substituted 2pyrrolidinones (2) since the N-protected γ -amino- α , β -unsaturated carboxylates (1) are readily prepared from naturally occuring α -amino acids.



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REFERENCES

- 1. Burgess, L. E.; Meyers, A. I. J. Org. Chem. 1992, 57, 1656-1662; and references cited therein.
- 2. Meyers, A. I.; Burgess, L. E. J. Org. Chem. 1991, 56, 2294-2296.
- 3. Burgess, L. E.; Meyers, A. I. J. Am. Chem. Soc. 1991, 113, 9858-9859.
- 4. Silverman, R. B.; Nanavati, S. M. J. Med. Chem. 1990, 33, 931-936.
- 5. Georgiadis, M. P.; Haroutounian, S. A.; Apostolopoulos, C. D. Synthesis. 1991, 379-381.
- 6. Knapp, S.; Gibson, F. S. J. Org. Chem. 1992, 57, 4802-4809; and references cited therein.
- 7. Takahata, H.; Takamatsu, T.; Yamazaki, T. J. Org. Chem. 1989, 54, 4812-4822; and references cited therein.
- 8. Pak, C. S.; Lee, E.; Lee, G. H. J. Org. Chem. 1993, 58, 1523-1530.
- 9. Hudlicky, T.; Sinai-Zingde, G.; Natchus, M. G. Tetrahedron Lett. 1987, 28, 5287-5290.
- 10. The structures assigned to products 2 are in agreement with their ¹H and ¹³C nmr spectral data, microanalysis (C,H,N) for (R)-2a-c and (S)-2c, and the high resolution mass spectrum for (S)-2d-e.
- 11. $[\alpha]_D^{23}$ = -13.9° (c 0.87, CHCl₃); Lit. $[\alpha]_D^{20}$ = -13.4° (c 0.87, CHCl₃), Craven, A. P.; Dyke, H. J.; Thomas, E. J. *Tetrahedron*, 1989, 45, 2417-2429.
- [α]_D²³= -37.5° (c 10, EtOH); Lit. [α]_D²⁰= +21.1° (c 0.54, EtOH) for the (S)-(+)-enantiomer, Nagao,
 Y.; Dai, W. M.; Ochiai, M.; Tsukagoshi, S.; Fujita, E. J. Org. Chem. 1990, 55, 1148-1156.
- 13. Pirkle, W. H.; Robertson, M. R.; Hyun, M. H. J. Org. Chem. 1984, 49, 2433-2437.